

| Syllabus – EAID CAA Placement Test Chemistry | |
|---|---|
| Topics: | Chemistry |
| Topics. | |
| | Matter and its properties |
| 1 | Chemistry and its scope |
| | • Units of measurement and conversion between them |
| | Sources errors and uncertainty in measurements |
| | Classification of matter, Changes of matter |
| | • Atomic theories, Atomic structure, Atomic spectra and their applications, Atomic composition |
| | Periodic table and how elements properties determined based on their location, Periodicity |
| | • Volume, temperature, pressure, and amount of a gas, Relationships among the four quantities of a gas and their calculations |
| | Characteristics of solutions and factors affecting solubility |
| | • Properties of solutions (qualitatively and quantitatively) |
| | • Electronic composition of the carbon atom |
| | • Diversity of organic compounds in terms of shape, size, and chemical and physical properties |
| | Classifications of organic compounds in terms of functional groups |
| | • Types of organic reactions and their applications |
| | Energy, force and conservation |
| 2 | Ionic, polar, and nonpolar covalent bonds, Shapes of molecules |
| | The concept of the mole and its applications (stoichiometry) |
| | Percent composition of a compound, Empirical and molecular formulas of a |
| | compound, Percent yield |
| | Acids and bases (strong and weak), The concept and use of pH scale |
| | • The concept of neutralization, Common ion effect, buffer solutions, and solubility |
| | Meaning of oxidation and reduction, redox reactions, and activity series |
| | • Redox reactions to produce electricity and manufacture electrolytic and galvanic |
| | cells |
| | • Factors affecting the reaction rate |
| | Chemical equilibrium |
| | • Energy changes during chemical reactions and/or physical changes, Hess's law and |
| | how it can be used to predict the occurrence of the chemical reaction |